

EMEZ ///

Content:

- why imaging why 3D...
- Imaging modes...
- Imaging "space"...
- Imaging principle
- Waves or particle Duality
- Photon no mass high speed
- Electron mass and fast
- Imaging modalities e.g in LM and EM....& Applications
- Resolution....
- other Imaging modes..... X-ray Tomo, Imaging, PET, MRI...

- Question to audience:
- Did you use any imaging techniques?
- LM..., EM..., MRI..., PET..., X-ray....?
- Who uses Image processing?
- Who is "comfortable" with contrasting/ staining technique?
- Who is familiar with labeling techniques?
- Who can explain the duality of the probes?
- Who is familiar with the resolution range of any mode?
- Do you know the difference between frequency and wavelength...?

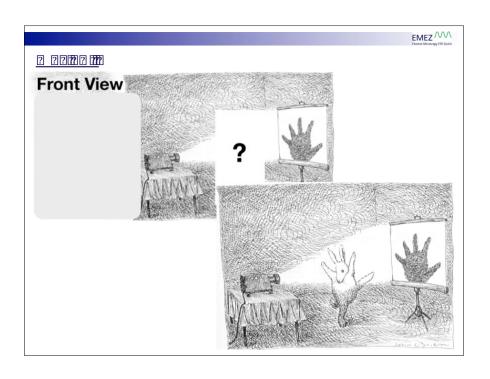
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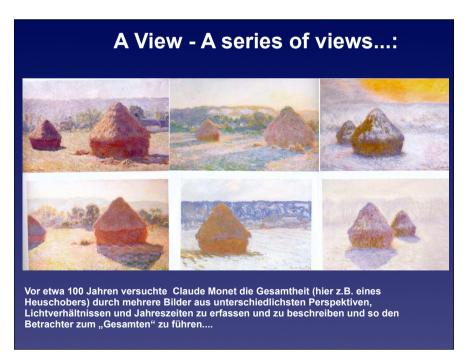
Aim of Imaging is primarily:

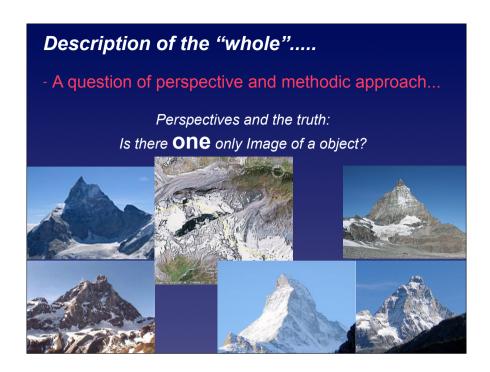
Get information on:

- Structure/ Morphology
- Chemical composition/molecular composition
- Regional function

As fast as possible, non-invasive and as natural as possible along all length scale to understand structure-function and time domain of life.

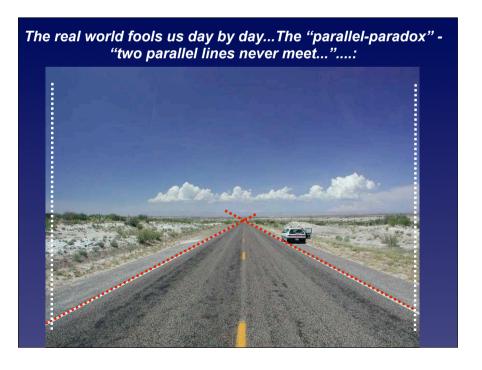


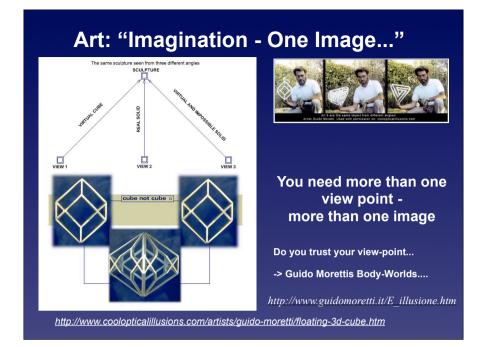












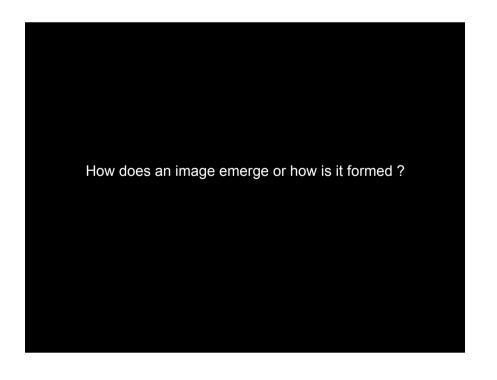
3D Imaging....Microscopy:

- Essential to make the "Nano-Word" -
 - "visible, tangible & concrete"

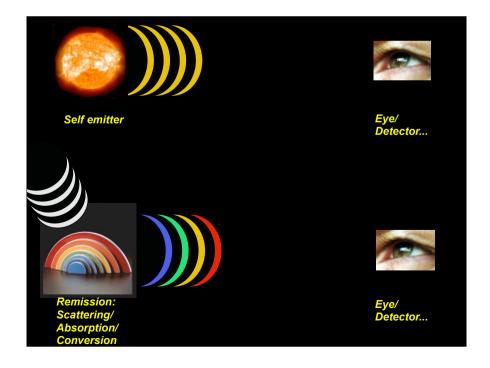
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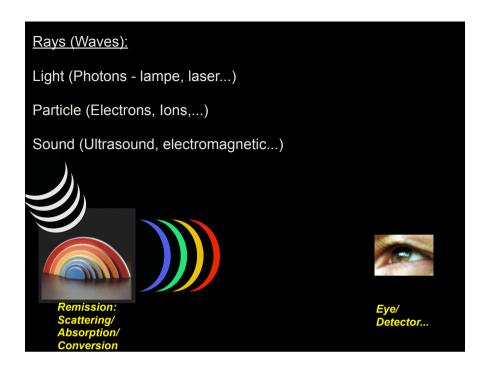
- 3D required because single images may miss-lead you.. (Viewpoint, Overlap, Irritation, Illusions, miss association....)
- 3D because the "small world" is translated to our "Thinking-Dimension"...you can turn it and hold it...
 - Visual perception and brain power.....

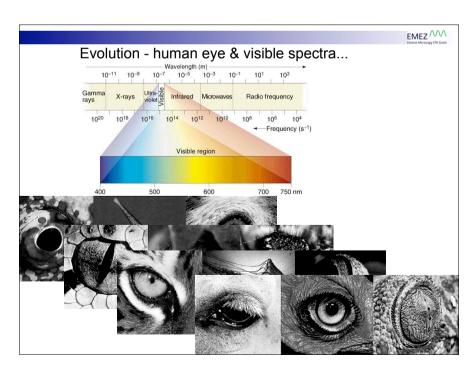
		EMEZ / CALL Electron Microscopy ETH Zurich
A virtual ex 1 Million tin	-	t's enlarge the world by
Than the real s	size of X becomes	s experienceable in our world
What:	Real Size:	Corresponding Size:
Atom	0,1-0,3nm	0,1-0,3mm (diameter body hair)
Protein	2-5nm	2-5mm (Rice corn)
Zytoskelett Fibers	9-25nm	9-25mm (robe)
Ribosom	25nm	25mm (2€ piece)
Nanoparticle	30-200nm	30-200mm (Walnut - CD)
		13

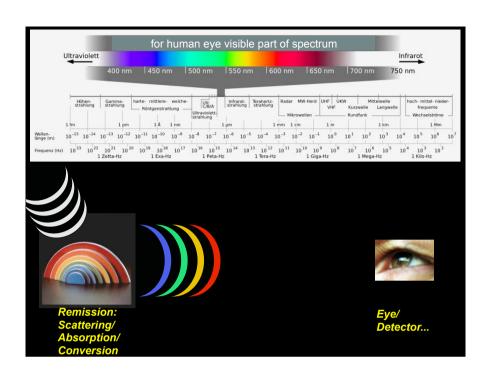


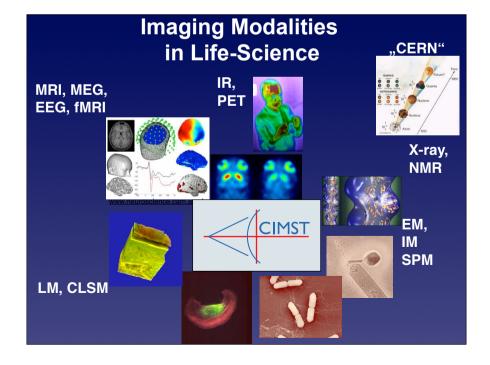


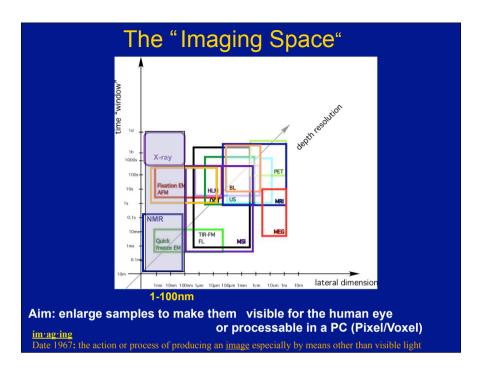


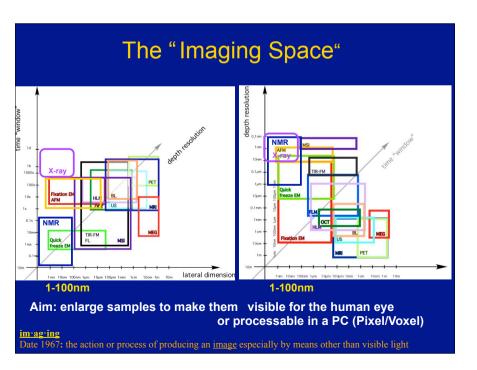


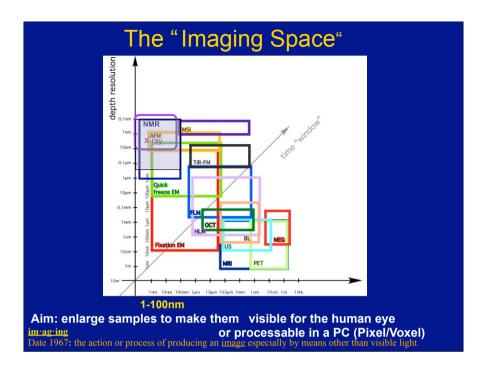


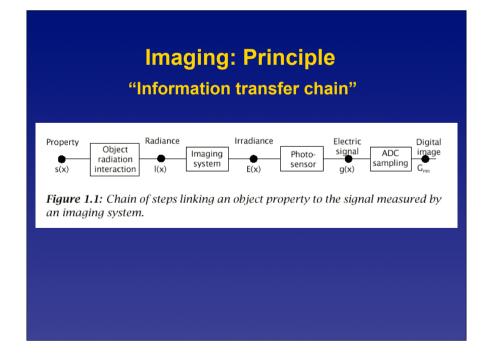














What is Electromagnetic Radiation?

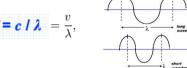
Electromagnetic radiation is energy – we describe it as a wave – visible light is only a small

The characteristics which distinguish different types of light are the electron-

- wavelength λ - frequency f

- energy

frequency

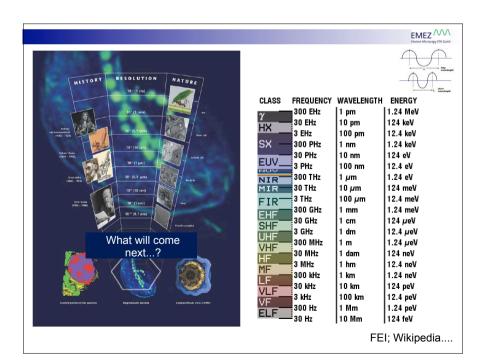


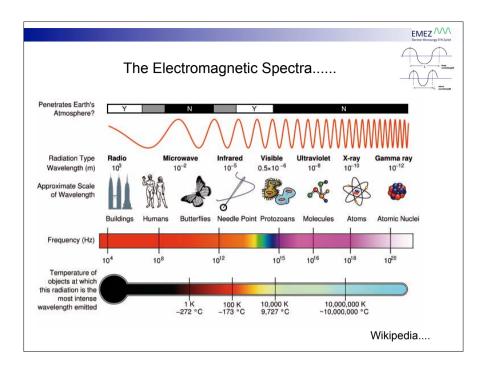
Wavelength: - the distance between two peaks (or two troughs) of the wave. Frequency: - the number of wavelengths passing a given point in one second.

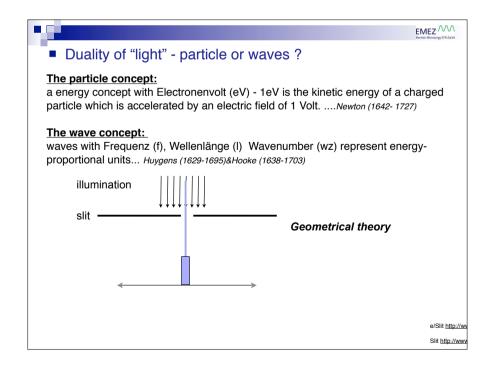
->The longer the wavelength, the more time it takes for a full wave to pass a given point (or the fewer waves pass the point in a given time).

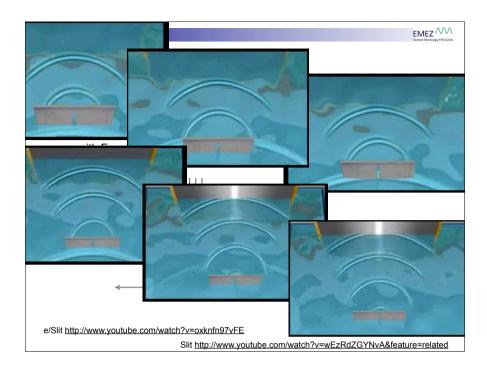
Since frequency is wavelengths per second, as the wavelength becomes longer the frequency decreases, and vice versa.

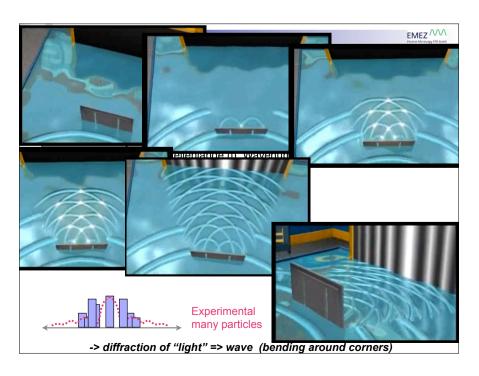
Energy: - is directly proportional to the frequency-- if the frequency increases, so does the energy of the radiation and vice versa.

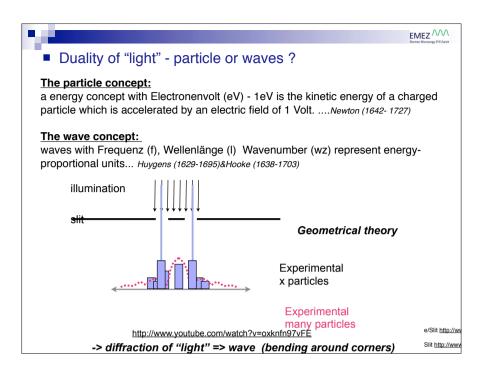
















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■ Duality of "light" - particle or waves.....

The particle concept:

a energy concept with Electronenvolt (eV) - 1eV is the kinetic energy of a charged particle which is accelerated by an electric field of 1 Volt.

-> "classical ray optics"

The wave concept:

waves with frequency (f), wavelength (λ) wave-number (wz) represent energy-proportional units.....

-> "wave optics" (wave front and Schrödinger eq.)

$$E = h \cdot f$$

(h...PLANCK's constant) directly relates the energy with the frequency of a "ray" Quantum Theory (Planck & Einstein)

The relation between the wavelength (λ) of a particle of mass, m, moving at a velocity, v, is given by the **DeBroglie** wave equation:

$$\lambda = \frac{h}{p}, \quad \lambda = \frac{h}{mv}$$
(1)



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■ Conversion of energy-> wavelength-> wave-number.....

$$E = h \cdot f$$

The wavelength of light is via the speed of light (c) linked to the frequency (general: $\lambda=\frac{v}{f}$,) => $\lambda=c$ / f

The so called wave-number is the reciprocal of the wavelength (wz, n) $n = 1 / \lambda$ (n Wave number usually in cm-1)

Conversion from one to the other unit:

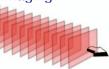
$$\lambda~[\mu m] = 10'000~/~n~[cm\text{-}1]$$
 ; $\lambda~[nm] = 10'000'000~/~n~[cm\text{-}1]$

$$E [eV] = 1 / 8065,5 * n [cm-1]$$

http://www.cactus2000.de/de/unit/masswav.shtm

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Not one wave but plane waves are used for imaging



In the physics of wave propagation, a **plane wave** is a constant-frequency wave whose wavefronts (surfaces of constant phase) are infinite parallel planes of constant amplitude normal to the phase velocity vector.

Mathematically, a plane wave is a wave of the following form:

$$u(\mathbf{x}, t) = Ae^{i(\mathbf{k} \cdot \mathbf{x} - \omega t)}$$

where *i* is the imaginary unit, **k** is the wave vector, ω is the angular frequency, and A is the (complex) amplitude.

-> two dimensional sinusoids have a frequency, phase, amplitude and direction!

(This becomes of interest when Fourier Transformations (FFT) are used to analyze images (Amplitude in real part, Phase in imaginary part))

Imaging: Principle

"Information transfer chain"

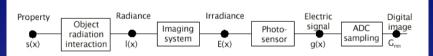
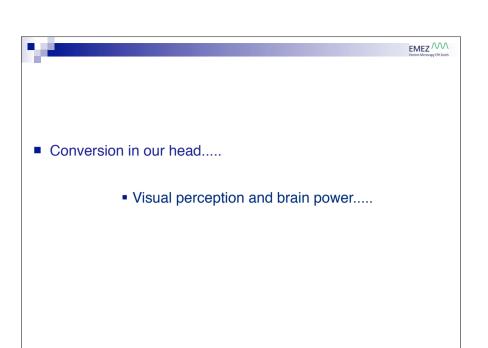


Figure 1.1: Chain of steps linking an object property to the signal measured by an imaging system.

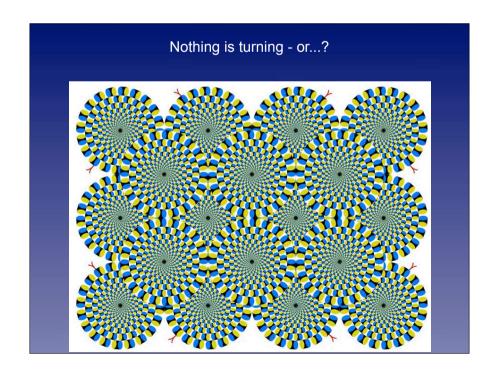
To image a certain structure the used wavelength should be in the range of the structure detail to be imaged... if not the wavelength can not interact linearly with the object...

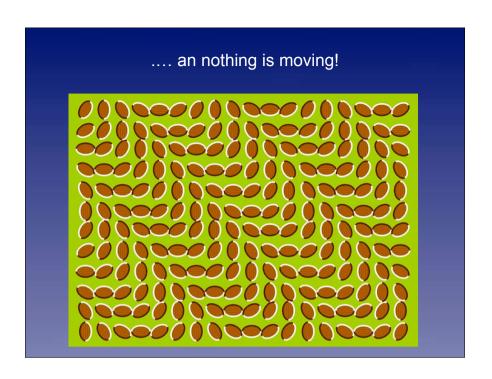


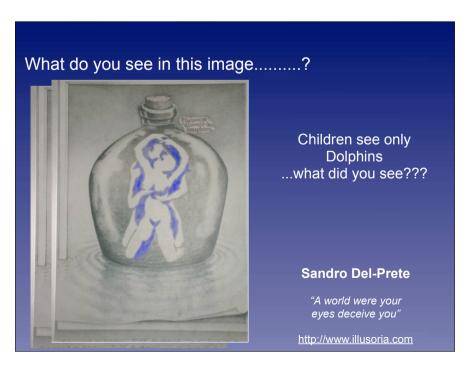
Can you trust what we see...

Watch out what your brain does! Just start to read (engl. or german)..... Aoccdrnig to a rscheearch at an Elingsh uinervtisy, it deosn't mttaer in waht oredr the ltteers in a wrod are, the olny iprmoetnt tihng is taht frist and lsat ltteer is at the rghit pclae. The rset can be a toatl mses and you can sitll raed it wouthit porbelm. Tihs is bcuseae we do not raed ervey lteter by it slef but the wrod as a wlohe. ceehiro. David Harris 2003: David R. Perrott and Callech senior research fellow Kourosh Saberi, Nature 1999 on speach "Cognitive Restoration of Reversed Speech." Afugrnud enier Sduite an enier Elingshcen Unvirestiät ist es eagl, in wlehcer Rienhnelfoge die Bcuhtsbaen in eniem Wrot sethen, das enizg wcihitge dbaei ist, dsas der estre und Izete Bcuhtsbae am rcihgiten

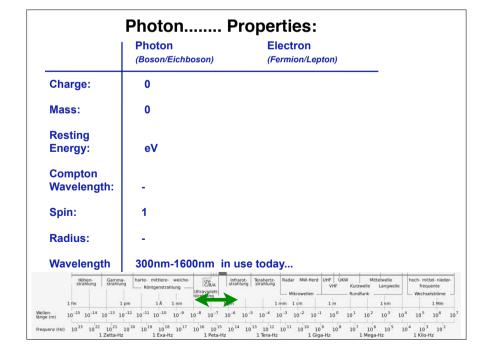
Paltz snid. Der Rset knan ttolaer Bölsdinn sien, und du knasnt es torztedm onhe Porbelme Iseen. Das ghet dseahlb, wiel wir nchit Bcuhtsbae für Bcuhtsbae enizlen Iseen, snodren Wröetr als Gnaezs.

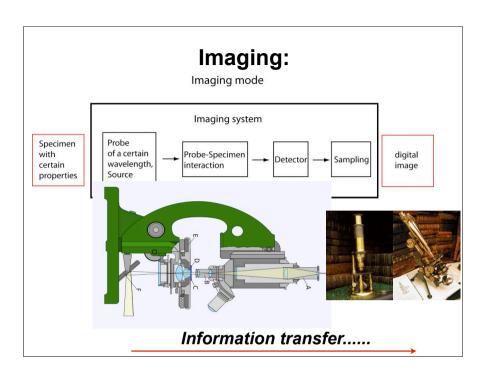


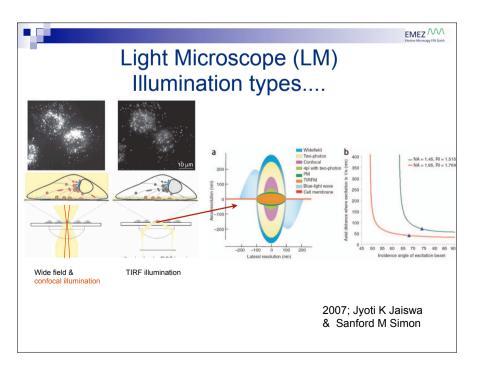




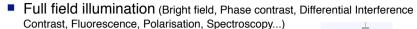








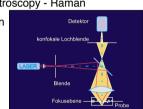


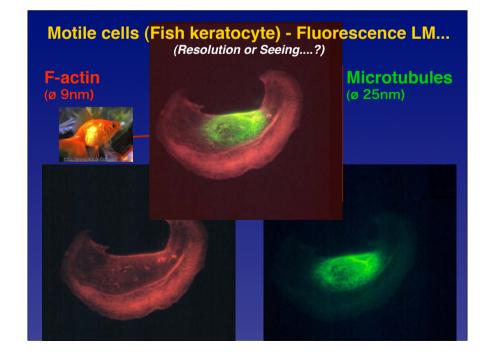


- -> parallel light is exposed to the whole specimen
- -> thickness of specimen is limiting resolution
- -> fast light exposure possible
- -> projection/reflexion images of the exposed area



- -> focused light spot is scanned through the specimen
- -> scanning in x,y and z possible
- -> thickness not so critical
- -> slow for large image area
- -> pixel by pixel images







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■ Imaging Mode: Light Microscopy (LM)

Probe: bundle of light or focused spot of light

• Wavelength: 300-1000nm

- Probe-Specimen Interaction: Absorption, elastic and inelastic scattering, conversion (fluorescence, Raman), phase shift...
- Modalities:
 - ☐ full-field illumination/Interference, Phase contrast, Scattering & Absorption, Emission
 - ☐ Scanning probe imaging/ Absorption, Emission, Scattering...
- Resolution:
- Application:
- Advantage:
- Limitation:

Electron: Energy & Wavelength

The dualism "wave - particle" is quantified by the De Broglie equation:

$$\lambda = h/p = h/mv$$

λ : wavelength; h: Planck constant; p: momentum

The energy of accelerated electrons is equal to their kinetic energy:

$$\mathsf{E} = \mathsf{eV} = \frac{\mathsf{m}_0 \mathsf{v}^2}{2}$$

V: acceleration voltage

 $e / m_0 / v$: charge / rest mass / velocity of the electron

$$p = m_0 v = (2m_0 eV)^{1/2}$$

$$\lambda = h / (2m_0 eV)^{1/2} (\approx 1.22 / V^{1/2} nm)$$

At the acceleration voltages used in TEM, relativistic effects have to be taken into account $\lambda = h / [2m_0eV (1 + eV/2 m_0/c^2)]^{-1/2}$

	Photon (Boson/Eichboson	Electron Properties: Electron (Fermion/Lepton)	
Charge:	0	-1e=-1,602x10 ⁻¹⁹ C	
Mass:	0	5,485x10 ⁻⁴ u (1 m _e) 9,109x10 ⁻³¹ kg	
Resting Energy:	eV	0,5109 MeV	
Compton Wavelength:	-	2,426x10 ⁻¹² m	
Spin:	1	+/- 1/2	
Radius:	-	2,817x10 ⁻¹⁵ m m _n /m _e = 1836,1	
Höhen- strahlung strahlu	Röntgenstrahlung Ultravio	W. Infrarot- Terahertz- Radar MW-Herd UHF UKW Mittelwele Strahlung Strahlung Strahlung Wick Hung UHF UKW Mittelwele Langwele Frequente Mitcrwellen Strahlung II m 1 m 1 km 1 km 1 km	

Wavelengths of Electrons

Accelerating voltages: SEM 0.5 – 30 kV TEM 100 – 1'000kV

V _{acc} / kV	Nonrelativistic wavelength [nm]	Relativistic wavelength [nm]
1	0.0388	0.0388
40	0.00613	0.0060
100	0.00386	0.00370
300	0.00223	0.00197
1000	0.00124	0.00087

(Atomic distances: ~ 0.1 nm (Å))

